STEREOSELECTIVE HYDROXYLATION OF N-CARBAMOYL-L-PYROGLUTAMATE. SYNTHESIS OF (-)-BULGECININE

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Abstract: Regio- and diastereoselective hydroxylation of the monoenolate derived from N-<u>t</u>-butoxycarbonyl-L-pyroglutamate (<u>1</u>) afforded the optically pure (4<u>R</u>)-hydroxypyroglutamate (<u>2a</u>) from which (-)-bulgecinine (<u>3</u>) was synthesized.

L-Pyroglutamic acid, a simple derivative of L-glutamic acid has been the efficient starting material for the synthesis of nitrogen containing natural products.^{1, 2)} Because of its bifunctionality, L-pyroglutamic acid has required the prior modification such as reduction^{1a)} to prevent the racemization at C-2 or thiolactam formation^{1b)} for the activation at C-5. Recent investigation on the reactivity towards nucleophilic reagents led us to find the direct chain elongation reaction at C-5 of N-carbamoylpyroglutamate.²⁾

As the succeeding study on the reactivity of the pyroglutamate, we wish to report that the regio- and diastereoselective hydroxylation of the monoenolate of N-<u>t</u>-butoxycarbonyl-L-pyroglutamate (<u>1</u>) affords the optically pure (4<u>R</u>)- hydroxypyroglutamate (<u>2a</u>) from which we have synthesized (-)-bulgecinine (<u>3</u>).³⁾

Our hydroxylation procedure is as follows. The pyroglutamate $(\underline{1}, 3.2 \text{ g}, 10 \text{ mmol})$ was treated with $\text{LiN}(\text{Me}_3\text{Si})_2$, prepared from $\text{HN}(\text{Me}_3\text{Si})_2$ (2.1 ml, 10 mmol) and an equivalent of 15% n-BuLi (6.5 ml, 10 mmol), in THF at -78 °C and the resulting enolate was oxidized with 2-toluenesulfonyl-3-phenyloxazilidine (2.7 g, 15 mmol) to give the 4-hydroxypyroglutamate⁴) [2a, mp 108-109 °C, $[\alpha]_D^{25}$ +19.2° (c 2.0, CHCl₃), 2.04 g, 61%].⁵)



No diastereoisomer $(\underline{4})^{6}$ was detected on HPLC of the alcohol $(\underline{2a}).^{7}$ $(4\underline{R})$ -Configuration of the alcohol $(\underline{2a})$ was determined by the NMR analysis of alcohols $\underline{2a}$ and $\underline{4}$. Namely, desielded signals of 2- and 4-hydrogen (δ 4.61 and 4.41, respectively) of the alcohol $(\underline{2a})$ compared to those (δ_{H-2} 4.44 and δ_{H-4} 4.24) of the epimer ($\underline{4}$) clearly indicates the 2,4-<u>trans</u> relationship of the alcohol ($\underline{2a}$). No racemization during the hydroxylation was detected in the HPLC analysis of (+)-MTPA ester ($\underline{2b}$).⁸)

Although the effect of N-carbamoyl substituent is not known, it is proposed that <u>re</u>-attack of the oxazilidine may become predominant when the coordination occurs as illustrated in Fig. 1.⁹) Steric repulsion of the benzyloxycarbonyl substituent and the lone pair of the enolate nitrogen present on the <u>si</u>-side, to the reagent may also contribute to the stereoselectivity.

4-Hydroxypyroglutamate (<u>2a</u>) thus synthesized was found to be the effective synthon for hydroxypyrrolidine derivatives such as (-)-bulgecinine (<u>3</u>) which is the main component of bulgecines, the unique glycopeptides isolated from <u>Pseudomonas acidophila</u> and <u>P. mesoacidophila</u>.³⁾ According to the procedure we developed,²⁾ (<u>4S</u>)-benzoxypyroglutamate (<u>5</u>) was treated with vinylmagnesium bromide at -40 °C in THF to afford the enone (<u>6</u>). Reduction of the enone (<u>6</u>) with NaBH₄-CeCl₃¹⁰ at -20 °C gave the alcohol (<u>7</u>) as an oil whose configuration at C-5 was not assigned.¹¹)



(Hypothetical Transition State Model)

Scheme 1

Fig. 1





Crude mesylate derived from the alcohol $\underline{7}$ on the treatment with MsCl/Et₃N in CH₂Cl₂ at 0 °C, did not change with Et₃N treatment, but rearranged to the 5-vinylprolinate ($\underline{8}$) when the crude mesylate was adsorbed on silica gel.¹²) The cyclized product ($\underline{8}$) was slowly eluted with 20% EtOAc in n-hexane. The fact that 2- and 5-hydrogens of the vinylprolinate ($\underline{8}$) showed two sets of NMR signals, δ 4.58 and 4.62 for H-2, and δ 4.55 and 4.62 for H-5, is characteristic for the 2,5-<u>trans</u> substituted N-carbamoylpyrrolidine derivatives.¹³) Moreover, the observed small coupling constant between 4- and 5-hydrogens, 0.7 or 1.1 Hz, clearly indicates the <u>trans</u> relationship of those hydrogens (Fig. 2).¹⁴)

Vinylprolinate <u>8</u> was then transformed to the alcohol (<u>9</u>) by ozonolysis at -78 ^OC and reduction with NaBH₄. Hydrolysis of the ester (<u>9</u>) followed by deprotection with trifluoroacetic acid gave the amino acid <u>3</u> [mp 195-196 ^OC, $[\alpha]_D^{29}$ -14.1^O (c 0.80, H₂O)] which showed identical spectroscopic data and physical constants with the natural bulgecinine³) in all respect.

The 4-hydroxypyroglutamate (<u>2a</u>) we synthesized may be regarded as an effective chiron for the synthesis of other useful nitrogen containing natural products.

<u>Acknowledgment</u>: The authors are grateful to Dr. Susumu Shinagawa, Takeda chemical Industries, Ltd., for his kind donation of natural bulgecinine and its spectral data. References and Notes

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- 4. $\delta_{\rm H}$ (CDCl₃, 100 MHz): 1.40 (9H, s), 4.41 (1H, dd, J=8, 10 Hz), 4.61 (1H, dd, J=2, 9 Hz), 5.19 (2H, s), 7.30 (5H, s).
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- 6. $\delta_{\rm H}$ (CDCl₃, 100 MHz): 1.42 (9H, s), 4.24 (1H, dd, J=4, 6 Hz), 4.44 (1H, dd, J=8, 8 Hz), 5.18 (2H, s), 7.32 (5H, s).
- 7. Column: 5 nm Lichrosorb SI-60, 4 mm id x 250 mm. Eluent: EtOAc-hexane (1:1), 2 ml/min. Detection at 254 nm. t_R : <u>2a</u>, 4.0 min; <u>4</u>, 5.9 min.
- 8. Column: 5 nm Lichrosorb SI-60, 4 mm id x 250 mm. Eluent: EtOAc-hexane (3:17), 1ml/min. Detection at 254 nm. t_R : <u>2b</u>, 8.2 min; (<u>+</u>)-<u>2a</u> (+)-MTPA ester, 8.2, 9.8 min.
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- 11. Although complex signals of carbinyl hydrogens (δ around 4.3 due to H-2 and H-5, and 5.2 due to H-4) made it difficult to assign the configuration of 5-hydroxyl, the simple splitting of 6-hydrogen signal (ddd at δ 5.86) implied that the alcohol ($\underline{7}$) was given with a high diastereomeric excess. $\underline{7}$, $\delta_{\rm H}$ (CDCl₃, 100 MHz): 1.35 (9H, s), 4.3 (2H, m), 5.11 (2H, s), 5.0-5.5 (4H, m), 5.86 (1H, ddd, J=5, 10, 17 Hz), 7.30 (5H, s), 7.3-7.6 (3H, m), 7.99 (2H, dd, J=2, 8 Hz).
- 12. Wakogel C-200, 100-200 mesh.
- 13. Unpublished results. Cf. reference 2.
- 14. Signal assignments were aided by H-H 2DJ, H-H and C-H COSY spectra at 500 or 125 MHz.

(Received in Japan 7 October 1987)